

Lafayette College
Department of Civil and Environmental Engineering

CE 321: Introduction to Environmental Engineering and Science

Fall 2019

Homework #11

Due: Wednesday, December 5, 2019

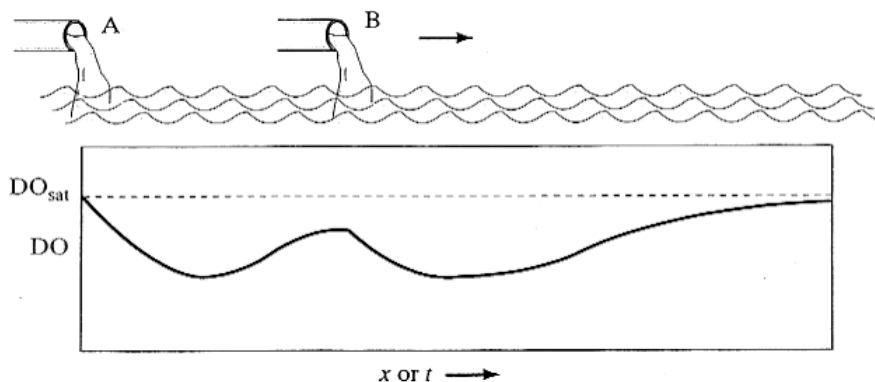
SOLUTIONS

- 1) Suppose some wastewater had a BOD_5 equal to a 180 mg/L and a reaction rate k equal to 0.22/day. It also has total Kjeldahl nitrogen content (TKN) of 30 mg/L.
 - a. Find the ultimate carbonaceous oxygen demand (CBOD).
 - b. Find the ultimate nitrogenous oxygen demand (NBOD).
 - c. Find the remaining BOD (nitrogenous plus carbonaceous) after five days have elapsed.

- 2) A wastewater treatment plant discharges 1.0 m³/s of effluent having an ultimate BOD of 40 mg/L into a stream flowing at 10.0 m³/s. Just upstream from the discharge point, the stream has an ultimate BOD of 3.0 mg/L. The deoxygenation constant k_d is estimated at 0.22/day.
 - a. Assuming complete and instantaneous mixing, find the ultimate BOD of the mixture of waste and river just downstream from the outfall.
 - b. Assuming a constant cross-sectional area for the stream equal to 55 m², what BOD would you expect to find at a point 10,000 m downstream.

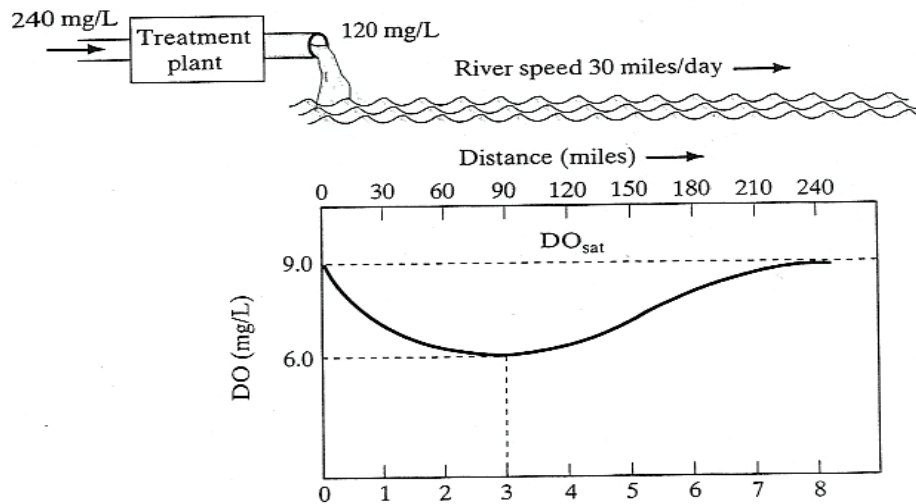
- 3) The wastewater in Problem 2 has DO equal to 4.0 mg/L when it is discharged. The river has its own DO, just upstream from the outfall, equal to 8.0 mg/L. Find the initial oxygen deficit of the mixture just downstream from the discharge point. The temperatures of sewage and river are both 15°C.

- 4) Two point sources of BOD along a river (A and B) cause the oxygen sag curve shown in the following image.



- a. Sketch the rate of reaeration vs. distance downriver.
- b. Sketch L_t (that is, the BOD remaining) as a function of distance downriver.

- 5) Untreated sewage with a BOD of 240 mg/L is sent to a wastewater treatment plant where 50 percent of the BOD is removed. The river receiving the effluent has the oxygen sag curve as shown in the following figure (the river has no other sources of BOD). Notice that downstream is expressed in both miles and days.



- Suppose the treatment plant breaks down and it no longer removes any BOD. Sketch the new oxygen sag curve starting just after the breakdown. Label the point which represents the critical distance downriver.
 - Sketch the oxygen sag curve, as it would have appeared four day after the breakdown of the treatment plant.
- 6) The ultimate BOD a river and sewage outfall, after mixing, is found to be 50 mg/L. Also the DO is found to be at a saturation value of 10.0 mg/L after mixing, therefore no initial deficit. The deoxygenation rate coefficient k_d is 0.30/day and the reaeration rate coefficient k_r is 0.90/day. The river is flowing at the speed of 48.0 miles per day. The only source of BOD in this river is the single outfall.
- Find the critical distance downstream at which DO is minimum.
 - Find the minimum DO.
 - If a wastewater treatment plant is to be build, what fraction of the BOD would have to be removed from the sewage to assure a minimum DO concentration of 5.0 mg/L everywhere downstream?
- 7) A city of 200,000 people deposits 37 cubic feet per second (cfs) of sewage having a BOD of 28.0 mg/L and 1.8 mg/L of DO into a river that has a flow rate of 250 cfs and a flow speed of 1.2 ft/s. Just upstream of the release point, the river has a BOD of 3.6 mg/L and a DO of 7.6 mg/L. The saturation value of DO is 8.5 mg/L. The deoxygenation coefficient k_d is 0.61/day and the reaeration coefficient k_r is 0.76/day. Assuming complete and instantaneous mixing of the sewage and river find
- The initial oxygen deficit and ultimate BOD just downstream of the outfall
 - The time and distance to reach the minimum DO
 - The minimum DO
 - The DO that could be expected 10 miles downstream

8) The town of Martins Creek, PA, has filed a complaint with the Department of Environmental Protection (DEP) citing the town of Portland, PA, for the discharge of raw sewage into the Delaware River. The raw sewage is considered to be the cause of high fecal coliform counts and reduced levels of dissolved oxygen (DO), which have lead to foul odors along the river between Portland and Martins Creek. The coliform counts and reduced DO levels have lead to restrictions of recreational areas within the Portland/Martins Creek reach of the Delaware River.

The DEP water quality criterion for the Delaware River is 5 mg/L of DO (i.e. at no point shall the DO concentration drop below 5 mg/L).

Martins Creek is 15.55 km down stream of Portland.

The following data pertain to the 7-year, 10-day low flow at Portland

Parameter	Wastewater	Delaware River just above Portland
Flow (m ³ /sec)	0.1507	1.08
BOD ₅ at 16°C (mg/L)	128.00	Not provided
BOD _u at 28°C (mg/L)	Not provided	11.40
DO (mg/L)	1.00	7.95
k at 20°C (day ⁻¹)	0.4375	k of BOD in river is based on WW
Temperature (°C)	16.00	28.00

Data of river after WWTP

Average Speed (m/sec)	N/A	0.390
Average Depth (m)	N/A	2.80
Bed-activity coefficient	N/A	0.200

A) What is the DO concentration as mg/L at Martins Creek?

WWTP

BOD₅ = 128 mg/L at 16 C

Flow = 0.1507 m³/sec

DO = 1.0 mg/L

T = 16 C

k = 0.4375



River before Portland

BOD_u = 11.40 mg/L

Flow = 1.08 m³/sec

DO = 7.95 mg/L

T = 28 C

River after Portland

Speed = 0.390 m/s

Depth = 2.8 m

Bed Activity = 0.20

Distance between Portland and Martins Creek is 15.55 km

B) Does your answer make sense based on the complaint received?

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1 GIVEN SOME WASTEWATER HAS A BOD_5 EQUAL TO 180 mg/L AND A REACTION RATE $k = 0.22/\text{day}$. IT ALSO HAS A TKNL OF 30 mg/L

FIND THE ULTIMATE CBOD
THE ULTIMATE NBOD
REMAINING TOTAL BOD AFTER 5 DAYS

ASSUMPTIONS CONSTANT TEMPERATURE

SOLUTIONS $CBOD_{ult} = C_0$ $180 \text{ mg/L} = L_0 (1 - e^{-5 \text{ days} \cdot 0.22/\text{day}})$ $L_0 = 269.81 \text{ mg/L}$

$NBOD_{ult} = 30 \text{ mg/L} \cdot 4.5714 \text{ mg O}_2/\text{mg N} = 137.1 \text{ mg O}_2/\text{L}$

$BOD \text{ REMAINING} = TBOD - BOD_5 = 269.8 \text{ mg O}_2/\text{L} + 137.1 \text{ mg O}_2/\text{L} - 180 \text{ mg O}_2/\text{L} = 226.9 \text{ mg/L O}_2$

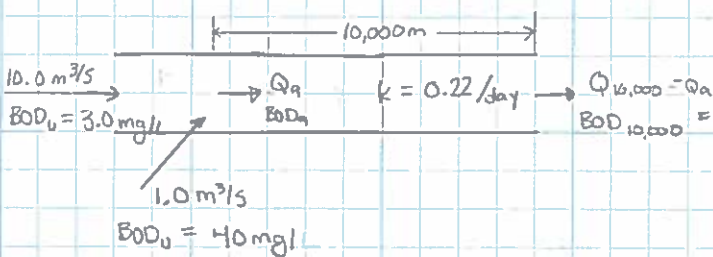
$CBOD_u = 269.8 \text{ mg/L O}_2$
 $NBOD_u = 137.1 \text{ mg/L O}_2$
 $BOD \text{ REMAINING}_5 = 226.9 \text{ mg/L O}_2$

2 GIVEN A WASTEWATER TREATMENT PLANT DISCHARGES $1.0 \text{ m}^3/\text{s}$ OF EFFLUENT W/ $BOD_u = 40 \text{ mg/L}$ INTO A STREAM W/ FLOW = $10.0 \text{ m}^3/\text{s}$. JUST UPSTREAM, OF THE DISCHARGE POINT, THE STREAM HAS $BOD_u = 3.0 \text{ mg/L}$. DEOXYGENATION CONSTANT $k_d = 0.22/\text{day}$. CROSS-SECTIONAL AREA OF STREAM = 55 m^2

FIND THE BOD_u OF THE MIX OF WASTEWATER AND RIVER JUST DOWNSTREAM FROM OUTFALL THE BOD AT A PT 10,000 m DOWNSTREAM

ASSUMPTIONS CONSTANT TEMP
INSTANTANEOUS, COMPLETE MIXING
BALANCE OF INS & OUTS (WATER & BOD)

SOLUTIONS



$Q_a = 10.0 \text{ m}^3/\text{s} + 1.0 \text{ m}^3/\text{s}$
 $10.0 \text{ m}^3/\text{s} \cdot 3.0 \text{ mg/L} + 1.0 \text{ m}^3/\text{s} \cdot 40 \text{ mg/L} = Q_a BOD_a$

$BOD_a = 6.3636 \text{ mg/L O}_2$

$V = \frac{11 \text{ m}^3/\text{s}}{55 \text{ m}^2} = 0.2 \text{ m/s}$

$\frac{10,000 \text{ m}}{0.2 \text{ m/s}} \cdot \frac{1 \text{ day}}{86,400 \text{ s}} = 0.578704 \text{ days}$

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2 (cont)

$$\text{BOD}_{10,000 \text{ m}} = 6.3636 \text{ mg/L} (1 - e^{-0.518704 \text{ day} \cdot 0.22 \text{ day}})$$

$$\text{BOD}_{10,000 \text{ m}} = \underline{0.77 \text{ mg/L O}_2}$$

$$\text{BOD}_a = 6.36 \text{ mg/L O}_2$$

$$\text{BOD}_{10,000 \text{ m}} = \underline{0.77 \text{ mg/L O}_2}$$

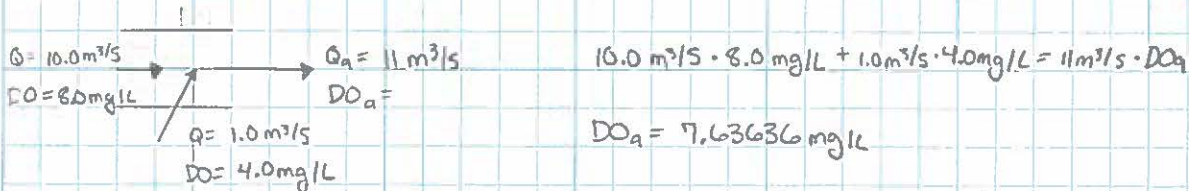
3 GIVEN THE WASTEWATER IN PROBLEM 2 HAS $\text{DO} = 4.0 \text{ mg/L}$ WHEN IT IS DISCHARGED. THE UPSTREAM DO OF THE RIVER IS 8.0 mg/L

FIND THE INITIAL OXYGEN DEFICIENCY OF THE MIX JUST DOWNSTREAM OF THE DISCHARGE POINT

ASSUMPTIONS

- CONSTANT TEMP OF RIVER & SEWAGE AT 15°C
- INSTANTANEOUS MIXING
- BALANCE OF BOD & WATER IN & OUT
- CHLORIDE CONCENTRATION = 0

SOLUTION

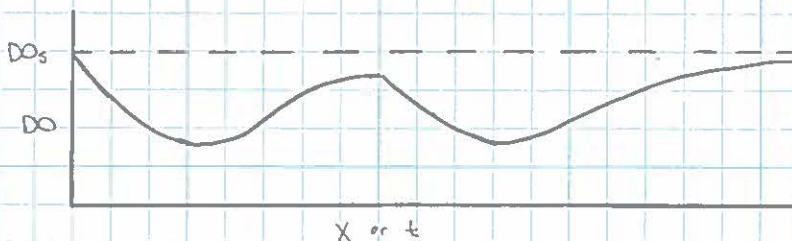


$$\text{DO}_a = 7.63636 \text{ mg/L}$$

$$D_a = \text{DO}_{S(15^\circ\text{C})} - \text{DO}_a = (10.08 - 7.63636) \text{ mg/L} = 2.443636 \text{ mg/L}$$

$$D_a = \underline{2.44 \text{ mg/L}}$$

4 GIVEN TWO POINT SOURCES OF BOD ALONG A RIVER CAUSE THE OXYGEN SAG CURVE BELOW



FIND SKETCH RATE OF REAERATION VS. DISTANCE DOWNSTREAM
SKETCH L_t AS A FUNCTION OF DISTANCE DOWNRIVER

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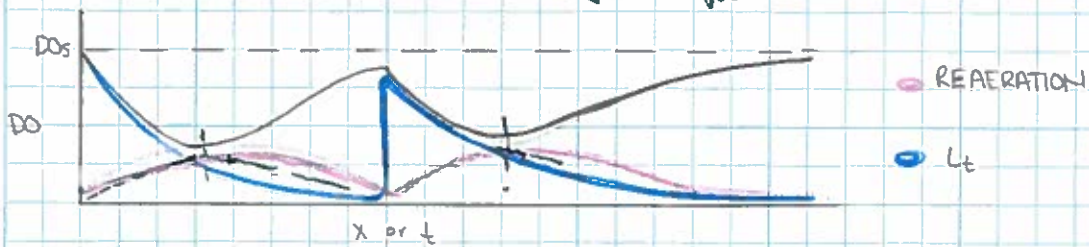
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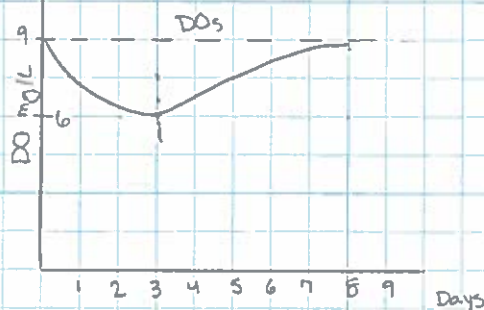
ASSUMPTIONS

CONSTANT TEMP
WELL MIXED
LITTLE TO NO BOD PRESENT AT START

SOLUTION



5 GIVEN 1 UNTREATED SEWAGE W/ BOD = 240 mg/L IS SENT TO A WASTEWATER TREATMENT PLANT WHERE 50% OF BOD IS REMOVED. THE RIVER RECEIVING THE EFFLUENT HAS NO OTHER SOURCES OF BOD & THE OXYGEN SAG CURVE SHOWN.
RIVER $v = 30$ mi/day



FIND THE NEW SAG CURVE IF THE TREATMENT PLANT WERE TO STOP FUNCTIONING & NO LONGER REMOVES BOD
THE SAG CURVE 4 DAYS AFTER BREAKDOWN OF THE PLANT

ASSUMPTIONS

WELL MIXED (INSTANTANEOUSLY)
PLUG FLOW
CONSTANT TEMPERATURE
CONSTANT RIVER SPEED

SOLUTION A- BOD $\times 2$ so $D \times 2$ SINCE $DO_s = DO_s$, t_c IS SAME

B- UNTIL DAY 4, WILL RESEMBLE GRAPH FROM PART A. AT 4 DAYS, JUMPS UP TO INITIAL SAG CURVE

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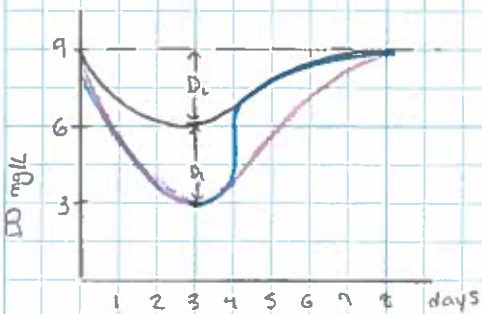
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5 (con't)



- INITIAL CURVE
- PART A - INFINITELY LONG AFTER BREAKDOWN
- PART B - 4 DAYS AFTER BREAKDOWN

6 GIVEN THE BOD_0 OF A RIVER JUST BELOW A SEWAGE OUTFALL = 50.0 mg/L AND THE $DO = DO_s = 10.0 \text{ mg/L O}_2$. THE DEOXYGENATION RATE COEFFICIENT $k_d = 0.30/\text{day}$ AND REGENERATION RATE $k_r = 0.90/\text{day}$. THE RIVER FLOWS AT 48.0 mi/day .

FIND CRITICAL DISTANCE DOWNSTREAM DISTANCE WHERE DO IS AT A MINIMUM.

DO MINIMUM

IF A WWTP IS TO BE BUILT, WHAT FRACTION OF BOD WOULD HAVE TO BE REMOVED FROM THE SEWAGE TO ASSURE DO IS ALWAYS $> 5.0 \text{ mg/L}$ ANYWHERE DOWNSTREAM

ASSUMPTIONS THE SEWAGE OUTFALL IS THE ONLY BOD SOURCE
 NO NET LOSS OR GAIN OF WATER
 EVEN, CONSISTENT MIXING
 CONSTANT TEMPERATURE
 CONSTANT RIVER SPEED

SOLUTION $k_d = 0.30/\text{day}$ $k_r = 0.90/\text{day}$ $D_0 = 0 \text{ mg/L}$

$$t_c = \frac{1}{0.90 - 0.30} \ln \left\{ \frac{0.90}{0.30} \left[1 - \frac{0(0.90 - 0.30)}{0.30 \cdot 50} \right] \right\} = 1.83102 \text{ days} \cdot \frac{48 \text{ mi}}{1 \text{ day}} = 87.889 \text{ mi}$$

$$DO_{1.83} = DO_s - D_{1.83} \quad DO_s = 10.0 \text{ mg/L}$$

$$D_{1.83} = \frac{0.30 \cdot 50}{0.90 - 0.30} (e^{-0.3 \cdot 1.83} - e^{-0.9 \cdot 1.83}) + 0e^{-0.90 \cdot 1.83} = 9.6225 \text{ mg/L O}_2$$

$$DO_{1.83} = 10.0 \text{ mg/L} - 9.6225 \text{ mg/L} = 0.3775 \text{ mg/L}$$

t_c independent of BOD ADDED \therefore WE WANT $DO_{1.83} = 5 \text{ mg/L}$

$$D_{1.83} = 10 - 5 = 5 \text{ mg/L}$$

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6 (cont) $D_1 = 5 \text{ mg/L} = \frac{0.3 \cdot \text{BOD}}{0.9 - 0.3} (e^{-0.3 \cdot 1.83} - e^{-0.9 \cdot 1.83})$ $\text{BOD} = 25.98 \text{ mg/L}$

% REMOVED = $\frac{50 \text{ mg/L} - 25.98 \text{ mg/L}}{50 \text{ mg/L}} \cdot 100\% = 48.04\%$

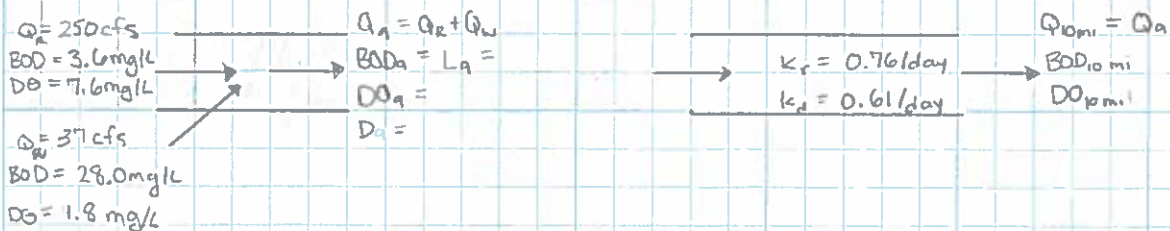
Dist_c = 87.9 mi
 DO_{1.83} = 0.377 mg/L O₂
 48.0% REDUCTION IN BOD REQUIRED

7 GIVEN A CITY OF 200,000 PEOPLE DEPOSITS 37 cfs OF SEWAGE OF SEWAGE WITH BOD = 28.0 mg/L AND 1.8 mg/L DO INTO A RIVER WITH A FLOW RATE OF 250 cfs AND A FLOW SPEED OF 1.2 ft/s. JUST UPSTREAM OF THE RELEASE POINT, THE RIVER HAS A BOD = 3.6 mg/L AND A DO = 7.6 mg/L. THE SATURATION VALUE OF DO = 8.5 mg/L. THE DEOXYGENATION COEFFICIENT $k_d = 0.61/\text{day}$ AND THE REGENERATION COEFFICIENT $k_r = 0.76/\text{day}$.

FIND INITIAL OXYGEN DEFICIT (D_1) AND ULTIMATE BOD JUST DOWNSTREAM OF OUTFALL
 THE TIME & DISTANCE TO REACH MINIMUM DO
 THE MINIMUM DO
 THE EXPECTED DO 10 mi DOWNSTREAM

ASSUMPTIONS COMPLETE, INSTANTANEOUS MIXING OF THE SEWAGE & RIVER
 CONSTANT TEMPERATURE
 CONSTANT RIVER SPEED & FLOW RATES
 NO OUTSIDE ADDITION/SUBTRACTION OF WATER OR BOD

SOLUTION



D_1 $D_1 = \frac{250 \text{ cfs} \cdot 7.6 \text{ mg/L} + 37 \text{ cfs} \cdot 28.0 \text{ mg/L}}{250 \text{ cfs} + 37 \text{ cfs}} = 6.85226 \text{ mg/L}$

$D_1 = DO_s - DO_1 = 8.5 \text{ mg/L} - 6.85 \text{ mg/L} = 1.65 \text{ mg/L O}_2$

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7 (cont) $L_a = \frac{3.4 \text{ mg/L} \cdot 250 \text{ cfs} + 37 \text{ cfs} \cdot 28.0 \text{ mg/L}}{250 \text{ cfs} + 37 \text{ cfs}} = 6.7456 \text{ mg/L}$

$$t_c = \frac{1}{0.76/\text{day} - 0.61/\text{day}} \ln \left[\frac{0.76/\text{day}}{0.61/\text{day}} \left[1 - \frac{1.6477 \text{ mg/L} (0.76/\text{day} - 0.61/\text{day})}{0.61/\text{day} \cdot 6.7456 \text{ mg/L}} \right] \right] = 1.052771 \text{ days}$$

$$t_c = 1.052771 \text{ days} \cdot \frac{86,400 \text{ s}}{1 \text{ day}} = 90959.4255 \text{ s} \cdot 1.2 \text{ f/s} = 109151.31 \text{ ft}$$

$$DO_{1.05d} = 8.5 \text{ mg/L} - D_{1.05d}$$

$$D_{1.05d} = \frac{0.61/\text{day} \cdot 6.7456 \text{ mg/L} (e^{-0.61/\text{day} \cdot 1.053 \text{ day}} - e^{-0.76/\text{day} \cdot 1.053 \text{ day}}) + 1.6477 \text{ mg/L} e^{-0.76/\text{day} \cdot 1.053 \text{ day}}$$

$$D_{1.05d} = 2.84859 \text{ mg/L} \quad DO_{\text{min}} = 8.5 \text{ mg/L} - 2.84859 \text{ mg/L} = 5.65141 \text{ mg/L}$$

$$\frac{10 \text{ mi}}{1.2 \text{ f/s}} \cdot \frac{5280 \text{ ft}}{1 \text{ mi}} \cdot \frac{1 \text{ d}}{86,400 \text{ s}} = 0.509259 \text{ d}$$

$$D_{0.509d} = \frac{0.61/\text{d} \cdot 6.7456 \text{ mg/L} (e^{-0.61/\text{d} \cdot 0.509 \text{ d}} - e^{-0.76/\text{d} \cdot 0.509 \text{ d}}) + 1.6477 \text{ mg/L} e^{-0.76/\text{d} \cdot 0.509 \text{ d}}$$

$$D_{0.509} = 2.5976788 \text{ mg/L}$$

$$DO_{10 \text{ mi}} = DO_s - D_{0.509} = 5.902 \text{ mg/L } O_2$$

$$\begin{aligned} D_a &= 1.65 \text{ mg/L} \\ L_a &= 6.75 \text{ mg/L} \\ t_c &= 1.05 \text{ days} \\ \text{Dist}_c &= 109151 \text{ ft} \\ DO_{\text{min}} &= 5.65 \text{ mg/L} \\ DO_{10 \text{ mi}} &= 5.90 \text{ mg/L} \end{aligned}$$

8 PROBLEM MARTIN'S CREEK PA HAS FILED A COMPLAINT CITING PORTLAND, PA FOR RELEASING UNTREATED SEWAGE INTO THE DELAWARE, WHICH THEY BELIEVE IS CAUSING LOW DO LEVELS & OTHER EFFECTS IN MARTIN'S CREEK.

GIVEN DEP CRITERION IS 5 mg/L DO. MARTIN'S CREEK IS 15.55 km DOWNSTREAM FROM PORTLAND. THE FOLLOWING DATA IS GIVEN FOR 7 yr 10 day FLOW AT PORTLAND.

	WASTEWATER	ABOVE PORTLAND DELAWARE RIVER	AFTER WWTP ENTERS RIVER
FLOW (m ³ /s)	0.1507	1.08	
BOD ₅ @ 16°C (mg/L)	128.00		AVG SPEED (m/s) 0.390
BOD ₅ @ 28°C (mg/L)		11.40	AVG DEPTH (m) 2.80
DO (mg/L)	1.00	7.95	BED ACTIVITY COEFFICIENT 0.200
K @ 20° (1/day)	0.4375	Based on ww	
TEMPERATURE (°C)	16.00	28.00	

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8 (cont) FIND DO AT MARTINS CREEK
DOES THIS MAKE SENSE BASED ON THE COMPLAINT

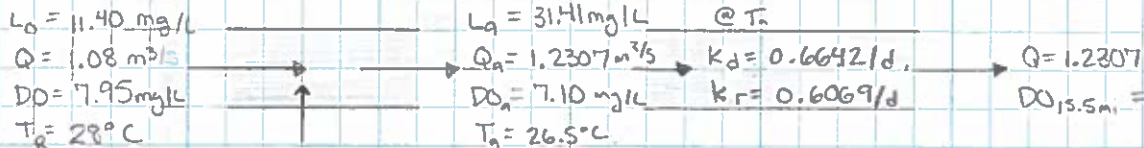
ASSUMPTIONS
INSTANTANEOUS, FULL MIXING
CONSTANT TEMP AFTER MIXING
K IN RIVER BASED ON K IN WW
PLUG FLOW AFTER MIXING
NO WATER OR BOD ENTERS OR LEAVES SYSTEM BESIDES WHAT IS SPECIFIED
STEADY STATE
NO CHLORIDE CONTENT

SOLUTION

$$k_{16} = 0.4375 \cdot 1.135^{(16-20)} = 0.26363 \text{ /day}$$

$$L_{0ww} (1 - e^{-5.6 \cdot 0.26363 \text{ /day}}) = 128 \text{ mg/L} \quad L_{0ww} = 174.775 \text{ mg/L}$$

RIVER



WWTP

$$L_0 = 174.775 \text{ mg/L}$$

$$Q = 0.1507 \text{ m}^3/\text{s}$$

$$DO = 1.0 \text{ mg/L}$$

$$T_w = 16^\circ\text{C}$$

$$k = 0.4375 \text{ (@ } 20^\circ\text{C) /day}$$

$$T_a = \frac{0.1507 \text{ m}^3/\text{s} \cdot 16^\circ\text{C} + 1.08 \text{ m}^3/\text{s} \cdot 28^\circ\text{C}}{1.2307 \text{ m}^3/\text{s}} = 26.5306^\circ\text{C} \quad k_{T_a} = 0.4375 \cdot 1.056^{(26-20)} = 0.62447846/\text{d}$$

$$DO_a = \frac{1.08 \text{ m}^3/\text{s} \cdot 7.95 \text{ mg/L} + 0.1507 \text{ m}^3/\text{s} \cdot 1.0 \text{ mg/L}}{1.2307 \text{ m}^3/\text{s}} = 7.09897 \text{ mg/L}$$

$$DO_{s_a} = \frac{30^\circ - 26.53^\circ}{30^\circ - 25^\circ} = \frac{7.56 \text{ mg/L} - DO_{s_a}}{7.56 \text{ mg/L} - 8.26 \text{ mg/L}} \quad DO_{s_a} = 8.0458$$

$$D_a = DO_{s_a} - DO_a = 0.94683$$

$$t_{MC} = \frac{15.55 \text{ km} \cdot 1000 \text{ m}}{0.39 \text{ m/s}} \cdot \frac{1 \text{ d}}{86400 \text{ s}} = 0.461479107$$

$$k_{d_{26}} = 0.4375 + \frac{0.390 \text{ m/s}}{2.90 \text{ m}} \cdot 0.200 = 0.465357/\text{day} \quad k_{d_{25}} = 0.465357 - 1.056^{(26.5-20)} = 0.66424/\text{d}$$

$$k_{r_{26}} = \frac{3.9 (0.340 \text{ m/s})^{1/2}}{(2.8 \text{ m})^{1/2}} = 0.51982823/\text{day} \quad k_{r_{25}} = 0.519828 \cdot 1.024^{(26.5-20)} = 0.606910/\text{d}$$

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8 (cont)

$$L_a = \frac{174.77 \text{ mg/L} \cdot 0.1507 \text{ m}^3/\text{s} + 1.08 \cdot 11.40 \text{ mg/L} \cdot \text{m}^3/\text{s}}{1.2307 \text{ m}^3/\text{s}} = 31.40538 \text{ mg/L}$$

$$D_{15,55 \text{ km}} = \frac{0.6642/\text{d} \cdot 31.41 \text{ mg/L}}{0.6069/\text{d} - 0.6642/\text{d}} (e^{-0.66 \cdot 0.4615} - e^{-0.607 \cdot 0.4615}) + 0.94683e^{-0.6069/\text{d} \cdot 0.4615 \text{ d}}$$

$$D_{15,55 \text{ km}} = 7.896067 \text{ mg/L}$$

$$DO_{mc} = 8.05 \text{ mg/L} - 7.90 \text{ mg/L} = 0.15 \text{ mg/L}$$

$$DO_{mc} = 0.15 \text{ mg/L}$$

THIS MAKES SENSE. A DO BELOW 5 IS UNDESIRABLE, AND THIS DO IS NEARLY ZERO. THIS WOULD RESULT IN A NEW TYPE OF ECOSYSTEM AND WOULD CAUSE THE COMPLAINTS OF MARTINL'S CREEK